



































Example

If the slope of the $(1/C_{dep})^2 vs. V_A$ characteristic is $2x10^{23}$ F⁻²V⁻¹, the intercept is 0.84V, and *A* is 1 μ m², find the lighter and heavier doping concentrations N_I and N_h .

Solution:

$$N_{l} = 2/(slope \times q\varepsilon_{s}A^{2})$$

$$= 2/(2 \times 10^{23} \times 1.6 \times 10^{-19} \times 12 \times 8.85 \times 10^{-14} \times 10^{-8} \text{ cm}^{2})$$

$$= 6 \times 10^{15} \text{ cm}^{-3}$$

$$V_{bi} = \frac{kT}{q} \ln \frac{N_{h}N_{l}}{n_{i}^{2}} \implies N_{h} = \frac{n_{i}^{2}}{N_{l}}e^{\frac{qV_{bi}}{kT}} = \frac{10^{20}}{6 \times 10^{15}}e^{\frac{0.84}{0.026}} = 1.8 \times 10^{18} \text{ cm}^{-3}$$
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